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15 Acids and Bases.

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strong acid. strong
base. weak acid.
weak base. an acid
that ionizes

completely in solvent.

a base that ionizes

completely in a

solvent. an acid that

releases few

hydrogen ions in

aqueous solution. a

base that

Chapter 15 Review

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**Acids Bases Modern
Chemistry**

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Chapter 14 Review

Acids Bases Bases. A chemical compound that increases the concentration of hydrogen ions in aqueous solution. Arrhenius acid. A substance that increases the concentration of

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hydroxide ions in
aqueous solution.

Arrhenius base. Acid
molecules attract

water molecules
which take the.

Chemistry Chapter 14

~Acids and Bases

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Chapter 14 Acids &
Bases Section 1

Properties of Acids &

Bases Aqueous

solutions of acids

taste sour. Acids

change the color of

acid-base indicators.

Some acids react with

active metals and

release hydrogen gas.

Acids react with

bases to produce

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salts and water. Acids
conduct electricity
(electrolytes).

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Chapter 14 & 15.

binary acid. oxyacid.

strong acid. weak

acid. an acid that

does not contain

oxygen, such as

hydrofluoric acid. an

acid that is a

compound of

hydrogen, oxygen, and

a third ele.... an acid

that ionizes

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Chemistry
completely in a
solvent.

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Chapter 14 Answers

Define a Lewis acid.

Can H^+ function as a Lewis acid? Explain your answer. A Lewis acid is a species that accepts an electron pair to form a covalent bond. H^+ is a Lewis acid because it can

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accept an electron pair from a base. For example, 3. Identify the Brønsted-Lowry acid and the Brønsted-Lowry base on the reactant side of each of the

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- Exercises - Strong
Acids, Strong Bases,
and pH - Page 738 9
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Answers react with active metals and release hydrogen gas.

4. Acids react with bases to produce salts and water. 5.

Acids conduct electric current. Modern

Chemistry: Acids and Bases (chapter 14)

Flashcards ... In

modern chemistry, we have a sound

understanding of

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Chemistry
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And Bases Test

**Modern Chemistry
Acids And Bases
Review Answers**

Acids react with bases to produce salts and water. 5. Acids conduct electric current. Binary Acid. An acid that contains only two different

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elements: hydrogen
and one of the more
electronegative
elements. Oxyacid.

An acid that is a
compound of
hydrogen, oxygen,
and a third element,
usually a nonmetal.

Modern Chemistry
Chapter 14: Acids
and Bases
Flashcards ...

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Chemistry Chapter 14

1 Modern

Chemistry Chapter 14.

Acids Bases ; 2

Section 1 Properties of
Acids Bases.

Aqueous solutions of
acids taste sour.

Acids change the
color of acid-base
indicators. Some
acids react with active
metals and release

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hydrogen gas. Acids react with bases to produce salts and water. Acids conduct electricity (electrolytes). 3

**PPT – Modern
Chemistry Chapter
14 PowerPoint
presentation ...**

Chapter 15 Acids and Bases. strong acid. strong base. weak

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Chapter Acids
And Bases Test

acid. weak base. an acid that ionizes completely in solvent. a base that ionizes completely in a solvent. an acid that releases few hydrogen ions in aqueous solution. a base that releases few hydroxide ions in aqueous solution.

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Acids Bases Modern Chemistry

an acid that does not contain oxygen, such as hydrofluoric acid.

oxyacid. an acid that is a compound of hydrogen, oxygen, and a third element, usually a nonmetal.

arrhenius acid. a substance that increases the concentration of

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hydronium ions in

aqueous solution.

strong acid.

And Bases Test

Long considered the
standard for honors

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(Chapter 17), and

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atomic and molecular

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biochemistry, and

medicine deepen

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