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temperature is normal at 98.6 0 F. What is

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the corresponding temperature on the
Celsius scale? $0\ 0\ 0\ 5\ 5\ 9\ 9\ (32)\ (98.6\ 32)$
 $)\ C\ F\ t\ t; t\ C = 37\ 0\ C\ 16-2$. The boiling
point of sulfur is $444.5\ 0\ C$. anschp16 -
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Expansion is related to the kinetic activity of the molecules, whereas expansion and phase changes of substances are more related to potential energy. $\frac{1}{2}mv^2$ T N Although not true in all cases, a good beginning is to define temperature as the average kinetic energy per molecule. Temperature vs. Internal

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Q.76GP. If heat is transferred to 150 g of water at a constant rate for 2.5 min, its temperature increases by 13 C° . When heat is transferred at the same rate for the

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same amount of time to a 150-g object of unknown material, its temperature increases by 61 C° .

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temperature is normal at 98.6 0 F. What is

the corresponding temperature on the
Celsius scale? 0 0 0 5 5 9 9 (32) (98.6 32)

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$C = \frac{5}{9}(F - 32)$; $t_C = 37.0\text{ }^\circ\text{C}$ 16-2. The boiling point of sulfur is $444.5\text{ }^\circ\text{C}$.

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16. TEMPERATURE 16.1. Thermal
Equilibrium Thermodynamics deals with

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the internal energies of systems and is governed by a set of laws (similar to Newton's law for mechanics). The central concept of thermodynamics is the temperature T . Properties of many bodies change as their thermal environment is altered. When the temperature increases, the volume of a liquid increases, the length

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of a metal rod increases, the electrical resistance increases, the pressure of a confined gas increases, etc.

16. TEMPERATURE

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Expansion and Expansion 16-1. Body temperature is normal at 98.6 °F. What is the corresponding temperature on the Celsius scale? $t_C = \frac{5}{9}(t_F - 32)$ (98.6 - 32) = 37 °C

16-2. The boiling point of sulfur is 444.5 °C.

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Figure 16.1 -The curvature of a bimetallic strip depends on temperature. (a) The strip is straight at the starting temperature, where its two components have the same length. (b) At a higher temperature, this strip bends to the right, because the metal

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on the left has expanded more than the metal on the right.

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STUDY. PLAY. Roger bacon. English philosopher and scientists of the 1200, a Franciscan monk who had studied at Oxford and Paris, viewed as a leading scholar of his time, known as Doctor Mirabilis wonderful teacher. Scientific Revolution.

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Chapter 16 Heat. $10 \text{ g} \cdot 1 \text{ cal/g} \cdot 2 \text{ }^\circ\text{C} = 20$
 cal . $20 \text{ g} \cdot 1 \text{ cal/g} \cdot 2 \text{ }^\circ\text{C} = 40 \text{ cal}$. 200 cal
 $/(100 \text{ g} \cdot 2 \text{ }^\circ\text{C}) = 1 \text{ cal/}^\circ\text{C}$ 200 cal
 $/(1000 \text{ g} \cdot 2 \text{ }^\circ\text{C}) = 0.1 \text{ }^\circ\text{C}$ The
amount of heat needed to raise 10 g of

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water 2 degrees cen.... The amount of heat
needed to raise 20 g of water 2 degrees
cen....

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EXPANSION 407 21.1 Temperature

The quantity that tells how hot or cold something is compared with a standard is temperature. We express temperature by a number that corresponds to a degree mark on some chosen scale.

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TEMPERATURE, HEAT, AND
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Thermal Contraction & Expansion.

Thermal Expansion- the volume of a material increases when a temperature increases. Remember Charles's Law? (As temperature increases, volume increases)

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Particles speed up, and have more collisions, and which makes even more collisions, and produce more force.

Thermal expansion/contraction are used in lots of things!

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Temperature is a fundamental quantity which characterizes the physical state of a substance. In the microscopic statistical theory, we understand temperature as the average energy per degree of freedom of motion of the substance.

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Thermodynamics Definition of heat: Heat
is the energy transferred between objects
... Thermal Expansion When the

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temperature of an object is raised, the
body usually exhibit “thermal expansion”.
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Expansion-particles of matter move farther

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apart to an increase in temperature ... the
WHAT its temperature rises when a given
amount of energy is absorbed by a given
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